

## CHEMICAL ENGINEERING THERMODYNAMICS-I

**Course Code:15CH1102**

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### Course Outcomes:

On successful completion of the course, the student should be able to

- CO 1** Discuss various thermodynamic processes.
- CO 2** Apply cubic equation of state to evaluate thermodynamic property with or without experimental data.
- CO 3** Recognize the need and state second law of thermodynamics. Analyze the second law from the study of heat engines.
- CO 4** Apply first and second law of thermodynamics to specific processes viz., pipe flow, nozzles, expansion and compression.
- CO 5** Compare and explain the different refrigeration and liquefaction processes and estimate the efficiency of the refrigeration system.

### UNIT-I

**(10 Lectures)**

#### INTRODUCTION:

The scope of thermodynamics, force, temperature, pressure, work, energy, heat.

#### THE FIRST LAW AND OTHER BASIC CONCEPTS:

Joule's Experiments, the first law of thermodynamics, thermodynamic state and path functions, enthalpy, the steady-state steady-flow process, equilibrium, the phase rule, the reversible process, heat capacity, constant-V and constant- P processes.

**UNIT-II****(10 Lectures)****VOLUMETRIC PROPERTIES OF PURE FLUIDS:**

The PVT behavior of pure substances, virial equations, the ideal gas, the applications of the virial equations, second virial coefficients from potential functions. Cubic equations of state, determination of Equation-of-State Parameters (The van der Waals and Redlich-Kwong equations of state only) generalized correlations for gases, generalized correlations for liquids.

**UNIT-III****(10 Lectures)****THE SECOND LAW OF THERMODYNAMICS:**

Statements of the second law, heat engines, thermodynamic temperatures scales, thermodynamic temperature and the ideal gas scale.

**ENTROPY:**

Entropy changes of an ideal gas, mathematical statement of the second law, the third law of thermodynamics, and entropy from the microscopic view point.

**UNIT-IV****(10 Lectures)****THERMODYNAMICS OF FLOW PROCESSES:**

Principles of conservation of mass, entropy and energy for flow systems, analysis of expansion processes; turbines, throttling; compression processes –compressors and pumps.

**REFRIGERATION AND LIQUEFACTION:**

The Carnot refrigerator, the vapor compression cycle, the comparison of refrigeration cycles, the choice of refrigerant, absorption refrigeration, the heat pump, liquefaction processes.

**UNIT-V****(10 Lectures)****THERMODYNAMIC PROPERTIES OF FLUIDS:**

Property relations for homogeneous phases, residual properties, two phase systems, thermodynamic diagrams, tables of thermodynamic properties.

**TEXT BOOK:**

Smith J.M. and Van Ness H.C, “*Introduction to Chemical Engineering Thermodynamics*”, 7<sup>th</sup> Edition, Tata McGraw Hill, 2009.

**REFERENCES:**

1. Rao Y.V. C., “*Chemical Engineering Thermodynamics*”, University Press Ltd., 2001.
2. Narayanan K. V., “*Chemical Engineering Thermodynamics*”, PHI, 2000.
3. Kyle B.G., “*Chemical and Process Thermodynamics*”, 3<sup>rd</sup> Edition, Pearson, Prentice Hall, 1999.
4. Abbott M.M, Van Ness H.C. “*Thermodynamics with chemical applications*”, 2<sup>nd</sup> Edition, Tata McGraw-Hill Publishing Company Limited, 2005.